

Enzymic synthesis of 3-[3-¹³C]dehydroquinic acidMartyn Frederickson,^a Emily J. Parker,^{a,b} John R. Coggins^c and Chris Abell^{*a}^a University Chemical Laboratory, Lensfield Road, Cambridge, UK CB2 1EW.

E-mail: ca26@cam.ac.uk; Fax: 44 1223 336362; Tel: 44 1223 336405

^b Centre for Structural Biology, Institute of Fundamental Sciences, Massey University, Palmerston North, New Zealand^c Division of Biochemistry and Molecular Biology, Institute of Biomedical and Life Sciences, University of Glasgow, Glasgow, UK G12 8QQ

Received 13th August 2003, Accepted 3rd September 2003

First published as an Advance Article on the web 9th September 2003

The title compound has been prepared enzymically over four steps from commercially available D-[5-¹³C]fructose.

Isotopically labelled compounds have been widely exploited in the field of biological chemistry. Molecules specifically labelled with either radioactive or nuclear active elements have allowed the elucidation of specific features of a plethora of enzyme mechanisms and have enabled researchers to determine the precise biochemical origins of particular chemical moieties within numerous biologically active molecules.

Dehydroquinase [3-dehydroquininate dehydratase; E.C. 4.2.1.10] is common to both the shikimate^{1,2} and quinate³ pathways. It catalyses the reversible interconversion of 3-dehydroquininate (DHQ) **1** and 3-dehydroshikimate (DHS) **2** (Fig. 1). The enzyme occurs in two chemically and biochemically distinct forms (type I and type II) that catalyse the same overall chemical transformation *via* different mechanisms. Type I enzymes are heat labile protein dimers⁴ that catalyse a *syn* elimination⁵ of water through Schiff's base intermediates⁶ involving a conserved lysine. Type II dehydroquinases are more thermally robust dodecameric species⁷ that effect an *anti* dehydration⁸ *via* an enzyme-stabilized enolate anion.

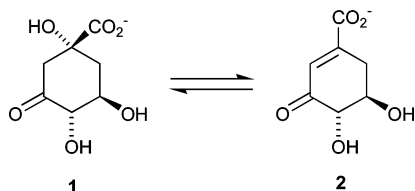


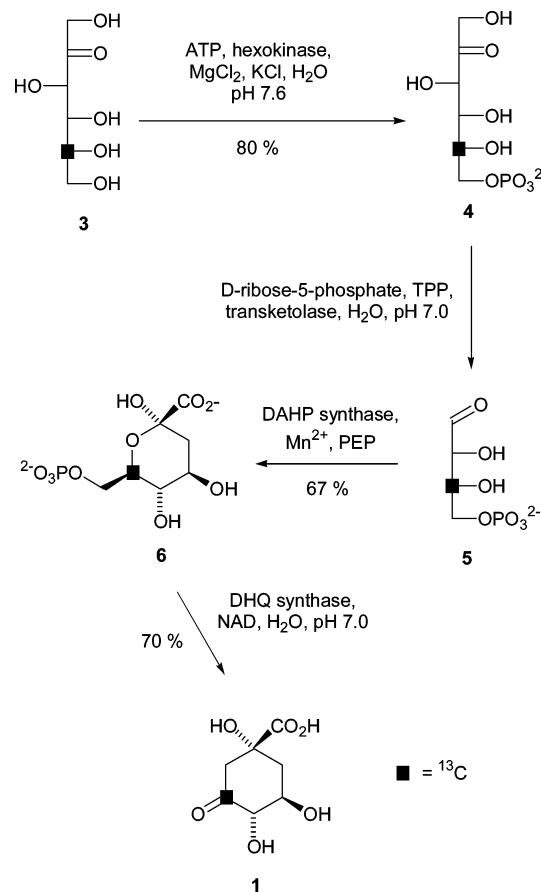
Fig. 1

As part of our long standing interest in the chemistry and biochemistry of dehydroquinase^{8–12} we required access to quantities of DHQ specifically labelled with ¹³C at the C(3) carbonyl. Chemical approaches to labelled DHQ using isotopically labelled cyanide or carbon monoxide were viable, however, we felt that synthetic routes were likely to be long and overly complicated by the protecting group manipulation that is so often necessary in the synthesis of sugars and related poly-hydroxylated carbocycles.

Enzymic synthesis, in contrast, appeared to allow a far more direct route to the desired labelled product. The high chemical specificity characteristic of enzymic systems often eliminates the need for protecting groups and thus seemed perfect for the synthesis of the small, highly functionalized water soluble target. As an added bonus, the enzymic synthesis also promised to deliver a series of highly functionalized and specifically labelled intermediates that could prove valuable for a variety of biochemical studies.

Herein we report the synthesis and purification of [3-¹³C]-DHQ over four enzyme-catalysed steps from a commercially

available isotopically labelled D-fructose precursor. Each of the four enzymic steps was high yielding (70–80% after purification). Key to the success of this approach was our ability to prepare and purify a series of labile phosphorylated intermediates *en route* to labelled DHQ. In order to minimize levels of decomposition, purification of all the phosphorylated intermediates was performed as rapidly as possible in a cold room at 4 °C.

Scheme 1 Enzymic synthesis of [3-¹³C] labelled DHQ.

We have previously described the use of shikimate pathway enzymes for the preparation of a series of specifically fluorinated^{13,14} and deuterated¹⁵ shikimate pathway intermediates. It therefore seemed reasonable to adopt a similar approach (utilizing the enzymes DAHP synthase and DHQ synthase) to make ¹³C labelled derivatives (Scheme 1). Central to this strategy was the need to prepare [3-¹³C] labelled D-erythrose-4-phosphate (E4P) **5**.

E4P has been prepared chemically by lead tetraacetate

oxidation of D-glucose-6-phosphate.¹⁶ It is notoriously difficult to work with, being both sensitive to pH and temperature and unstable with respect to phosphate elimination. In dilute aqueous solution it exists as a mixture of the monomeric aldehyde and its hydrate but at higher concentrations it rapidly converts to a complex mixture of cyclic dimeric forms^{17–19} which make chromatographic purification difficult. For these reasons we chose not to isolate **5** but to convert it directly to 3-deoxy-D-arabino-heptulosonate-7-phosphate (DAHP) **6** using DAHP synthase. Synthesis of **5** was achieved from commercially available D-[5-¹³C]fructose²⁰ **3** via a procedure involving chemospecific phosphorylation at C(6) followed by the transfer of a C₂ unit to D-ribose-5-phosphate (R5P).

The relatively high degree of substrate tolerance of hexokinase together with its ready commercial availability made it the enzyme of choice for the first step. Treatment of **3** (25 mg) with ATP and hexokinase at ambient temperature afforded D-[5-¹³C]fructose-6-phosphate **4**. The conversion was monitored by assaying small aliquots of the reaction mixture for the formation of ADP using a standard pyruvate kinase–lactate dehydrogenase coupled assay system.²¹ Spectrophotometric analysis (monitoring at 340 nm for the consumption of NADH) showed phosphorylation to be essentially complete after 1 hour. Purification by anion-exchange chromatography (Amersham MonoQ® eluting with an NH₄HCO₃ gradient) afforded **4** (80%) upon lyophilization.

To catalyse the conversion of **4** to E4P **5** we utilized transketolase, a thiamine pyrophosphate (TPP) dependent enzyme and coupled this to the phosphoenolpyruvate (PEP) utilizing enzyme DAHP synthase.²² Thus a solution of **4**, R5P, TPP, transketolase, PEP and DAHP synthase was held at ambient temperature and aliquots of the reaction mixture were analysed spectrophotometrically (monitoring at 260 nm for the consumption of PEP). After around 5 hours PEP consumption had ceased. Purification by anion-exchange chromatography (Amersham MonoQ® eluting with an NH₄HCO₃ gradient) and lyophilization afforded the desired phosphate **6** (67%), the overall yield representing an average of 82% per enzymic step which was acceptable given the instability of **5**.

Synthesis of **1** was readily completed from **5** using the NAD-dependent enzyme DHQ synthase.²³ A mixture of **6**, NAD and DHQ synthase was held at ambient temperature and assayed spectrophotometrically for the formation of **1** by monitoring the appearance of DHS **2** (234 nm) upon treatment of an aliquot of the reaction mixture with type II dehydroquinase.²⁴ Formation of **1** was found to be essentially complete within 3 hours. Purification by HPLC (BioRad Aminex® HPX-87H) eluting with 50 mM formic acid afforded 3-[3-¹³C]DHQ **1** (10 mg, 70%) as a colourless gum upon lyophilization. The product was shown to be essentially homogeneous by proton-decoupled ¹³C NMR spectroscopy; only a single resonance could be observed and was located in the carbonyl region of the spectrum (δ_c 209.5).

In order to determine the efficiency of the synthesis we followed the overall conversion of **3** to **1** by ¹³C NMR spectroscopy to trace the position of the ¹³C label during each of the enzyme catalysed steps. Thus a sample of **3** (in H₂O containing 10% ²H₂O) was treated sequentially with the four enzymes (in the presence of all of the necessary reagents and cofactors) and ¹³C NMR spectra recorded at each stage to monitor the resulting interconversion (Fig. 2).

D-[5-¹³C]Fructose **3** appears to exist in solution as a mixture of three interconverting isomeric species (Fig. 2a). Addition of hexokinase resulted in the rapid and relatively clean conversion to the 6-phosphate **4** which exists in solution as an isomeric pair of phosphorus coupled (J_{C-P} 8 Hz) species (Fig. 2b). Upon subsequent treatment with a mixture of transketolase and DAHP synthase there was a simplification of the reaction mixture and a phosphorus coupled (J_{C-P} 5 Hz) signal attri-

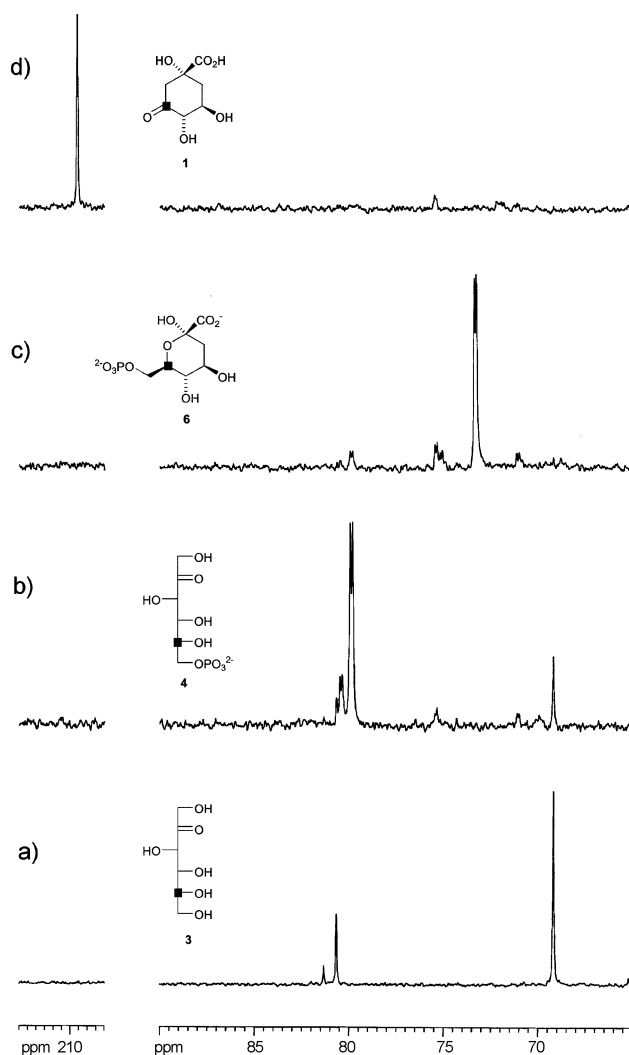


Fig. 2 Proton decoupled ¹³C NMR spectra (62.9 MHz) following the conversion of D-[5-¹³C]fructose **3** to 3-[3-¹³C]dehydroquinate **1**. a) D-[5-¹³C]fructose **3**; b) formation of D-[5-¹³C]fructose-6-phosphate **4** upon addition of hexokinase; c) formation of [6-¹³C]DAHP **6** upon addition of transketolase and DAHP synthase; d) formation of 3-[3-¹³C]DHQ **1** upon addition of DHQ synthase.

butable to DAHP **6** was observed. At this stage the mixture contained only trace quantities of the intermediates **4** and **5** and the starting material **3** had been completely consumed (Fig. 2c). After subsequent treatment with DHQ synthase we observed only a single resonance (Fig. 2d) in the ¹³C NMR spectrum corresponding to the carbonyl signal (δ_c 209.5) of 3-[3-¹³C]DHQ **1**.

In summary, we report a short and efficient synthesis of 3-[3-¹³C]DHQ **1** from commercially available D-[5-¹³C]fructose **3** over four enzyme-catalysed steps. The purified product has been shown to be essentially homogeneous with respect to ¹³C incorporation. Additionally we have used the ¹³C label in **3** to monitor the progress of each of the enzymic steps and have shown the overall conversion to be extremely ¹³C atom efficient.

We thank the BBSRC for postdoctoral support (to M.F. and E.J.P.).

Notes and references

- 1 E. Haslam, *Shikimic acid: Metabolism and Metabolites*, John Wiley and Sons, Chichester, 1993.
- 2 C. Abell, in *Comprehensive Natural Products Chemistry*, ed. U. Sankawa, Elsevier, Amsterdam, 1999, vol. 1, p. 573.
- 3 N. H. Giles, M. E. Case, J. A. Baum, R. F. Geever, L. Huiet, V. B. Patel and B. M. Tyler, *Microbiol. Rev.*, 1985, **49**, 338.

-
- 4 S. Chaudhuri, K. Duncan, L. D. Graham and J. R. Coggins, *Biochem. J.*, 1991, **275**, 1.
 - 5 B. W. Smith, M. J. Turner and E. Haslam, *J. Chem. Soc., Chem. Commun.*, 1970, 842.
 - 6 A. Schneier, C. Kleanthous, R. Deka, J. R. Coggins and C. Abell, *J. Am. Chem. Soc.*, 1991, **113**, 9416.
 - 7 A. R. Hawkins, W. R. Reinhert and N. H. Giles, *Biochem. J.*, 1982, **203**, 769.
 - 8 J. M. Harris, C. González-Bello, C. Kleanthous, A. R. Hawkins, J. R. Coggins and C. Abell, *Biochem. J.*, 1996, **319**, 333.
 - 9 M. Frederickson, E. J. Parker, A. R. Hawkins, J. R. Coggins and C. Abell, *J. Org. Chem.*, 1999, **64**, 2612.
 - 10 A. W. Roszak, D. A. Robinson, T. Krell, I. S. Hunter, M. Frederickson, C. Abell, J. R. Coggins and A. J. Laphorn, *Structure (London)*, 2002, **10**, 493.
 - 11 M. Frederickson, J. R. Coggins and C. Abell, *Chem. Commun.*, 2002, 1886.
 - 12 M. D. Toscano, M. Frederickson, D. P. Evans, J. R. Coggins, C. Abell and C. González-Bello, *Org. Biomol. Chem.*, 2003, **1**, 2075.
 - 13 P. J. Duggan, E. Parker, J. Coggins and C. Abell, *Bioorg. Med. Chem. Lett.*, 1995, **5**, 2347.
 - 14 S. Balasubramanian, G. M. Davies, J. R. Coggins and C. Abell, *J. Am. Chem. Soc.*, 1991, **113**, 8945.
 - 15 S. Balasubramanian and C. Abell, *Tetrahedron Lett.*, 1991, **32**, 963.
 - 16 A. S. Sieben, A. S. Perlin and F. J. Simpson, *Can. J. Biochem.*, 1966, **44**, 663.
 - 17 P. F. Blackmore, J. F. Williams and J. K. Macleod, *FEBS Lett.*, 1976, **64**, 222.
 - 18 J. F. Williams, P. F. Blackmore, C. C. Duke and J. K. Macleod, *Int. J. Biochem.*, 1980, **12**, 339.
 - 19 C. C. Duke, J. K. Macleod and J. K. Williams, *Carbohydr. Res.*, 1981, **95**, 1.
 - 20 D-[5-¹³C]fructose was obtained from Omicron Biochemicals, Inc., 1347 North Ironwood Drive, South Bend, IN 46615-3566, USA (product code: FRU-006).
 - 21 C. Blondin, L. Serina, L. Weismuller, A. M. Giles and O. Barzu, *Anal. Biochem.*, 1994, **220**, 219.
 - 22 C. M. Stephens and R. Bauerle, *J. Biol. Chem.*, 1991, **266**, 20810.
 - 23 J. W. Frost, J. L. Bender, J. T. Kadonaga and J. R. Knowles, *Biochemistry*, 1984, **23**, 4470.
 - 24 S. Chaudhuri, J. M. Lambert, L. A. McColl and J. R. Coggins, *Biochem. J.*, 1986, **239**, 699.